

# CBCS Scheme

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15AE33

Third Semester B.E. Degree Examination, June/July 2018

## Aero Thermodynamics

Time: 3 hrs.

Max. Marks: 80

- Note:** 1. Answer any FIVE full questions, choosing one full question from each module.  
2. Use of Thermodynamics data handbook is permitted.

### Module-1

- 1 a. Distinguish between : i) Intensive and Extensive properties ii) Diathermal and Adiabatic wall. (04 Marks)  
b. State Zeroth law of thermodynamics and extract the concept of temperature from it. (06 Marks)  
c. The temperature  $T$  on a thermometric scale is defined as  $T = a \ln k + b$ , where  $a$  and  $b$  are constants. The values of  $k$  are found to be 1.83 and 6.78 at  $0^\circ\text{C}$  and  $100^\circ\text{C}$  respectively. Calculate the temperature for a value of  $K = 2.42$ . (06 Marks)

OR

- 2 a. Bring out two similarities and two dissimilarities between heat and work. (04 Marks)  
b. Derive an expression for displacement work in a polytropic process  $pV^n = \text{constant}$ . Show on a  $p$ - $v$  diagram four expansion processes for  $n = 0$ ,  $n = 1$ ,  $n = 1.4$  and  $n = \infty$ . Name each of the process. (08 Marks)  
c. A shaft transmitting 600 hp rotates at 3600 rpm. Determine the torque applied to the shaft. (04 Marks)

### Module-2

- 3 a. Write the first law of thermodynamics for any process in :  
i) closed-system ii) - Open system. (04 Marks)  
b. A stationary mass of gas is compressed without friction from initial state of  $0.3\text{m}^3$  and  $0.105\text{MPa}$  to a final state of  $0.15\text{m}^3$  and  $0.105\text{MPa}$ , the pressure remaining constant during the process. There is a transfer of  $37.6\text{kJ}$  of heat from the gas during the process. How much does the internal energy of the gas change? (06 Marks)  
c. A domestic refrigerator is loaded with food and the door closed. During a certain period of time the machine consumes  $1\text{kWh}$  of energy and the energy of the system decreases by  $5000\text{kJ}$ . Determine the magnitude and direction of heat transfer for the process. (06 Marks)

OR

- 4 a. Write Steady Flow Energy Equation and explain all the terms involved. (04 Marks)  
b. Apply SFEE for : i) Adiabatic Nozzle ii) Steam turbine. (06 Marks)  
c. A small turbine runs an aircraft refrigeration system. Air enters the turbine at 4 bar and  $40^\circ\text{C}$  with velocity  $200\text{m/s}$ . If the work output of the turbine is  $52\text{kJ/kg}$  of air, calculate the heat transferred per kg of air. (06 Marks)

### Module-3

- 5 a. Represent schematically and give performance equation for : (04 Marks)  
i) Heat engine ii) Refrigerator iii) Heat pump.  
Prove that  $(\text{COP})_{\text{HP}} = (\text{COP})_{\text{Refrigerator}} + 1$ .  
b. State Kelvin Planck and Clausius statements of second law of thermodynamics and show that they are equivalent. (06 Marks)

- c. A reversible refrigerator operates between  $35^{\circ}\text{C}$  and  $-15^{\circ}\text{C}$ . If heat rejected to  $35^{\circ}\text{C}$  is  $1.5\text{kw}$ , determine the rate at which heat is leaking into refrigerator. (06 Marks)

OR

- 6 a. Define Entropy and prove that it is a property of the system. (04 Marks)  
 b. For an ideal gas undergoing finite change of state from 1 to 2, derive an expression for change in entropy. (05 Marks)  
 c. A block of iron weighing  $100\text{kg}$  and having a temperature of  $100^{\circ}\text{C}$  is immersed in  $50\text{kg}$  of water at a temperature of  $20^{\circ}\text{C}$ . What will be the change in entropy of the combined system of iron and water? Specific heats of iron and water are  $0.4\text{kJ/kg K}$  and  $4.18\text{ kJ/kg K}$  respectively. (07 Marks)

Module-4

- 7 a. Define : i) Pure substance ii) Saturation conditions iii) Triple point iv) Critical point v) Compressibility factors. (05 Marks)  
 b. Sketch and explain P – T diagram of water. (05 Marks)  
 c. Find enthalpy, entropy and volume of steam at  $1.4\text{MPa}$  and  $380^{\circ}\text{C}$ . (06 Marks)

OR

- 8 a. Derive and explain Maxwell's equations. (08 Marks)  
 b. Show that for an ideal gas  $C_p - C_v = R$ . (02 Marks)  
 c.  $1\text{ kg}$  of air at a pressure of  $8\text{ bar}$  and temperature  $100^{\circ}\text{C}$  undergoes reversible polytropic process following the law  $pv^{1.2} = \text{constant}$ . If final pressure is  $1.8\text{ bar}$  determine the final specific volume, Temperature and increase in entropy. Assume  $R = 0.287\text{ kJ/kg k}$ ,  $\gamma = 1.4$ . (06 Marks)

Module-5

- 9 a. What are Air standard Assumptions? (04 Marks)  
 b. Explain Working of a diesel engine with the help of p – v and T-S diagrams. Derive an expression for the efficiency of diesel cycle in terms of its compression and cut – off ratios. (07 Marks)  
 c. A diesel engine has a compression ratio of 14 and cut – off takes place at 6% of stroke. Find Air – standard efficiency. (05 Marks)

OR

- 10 a. Explain Rankine cycle with the help of a sketch and T – S diagram. Derive an expression for thermal efficiency of Rankine cycle. (06 Marks)  
 b. What are the methods for increasing the efficiency of Rankine cycle? (04 Marks)  
 c. Consider a steam power plant operating on a simple Rankine cycle. Steam enters the turbine at  $3\text{MPa}$  and  $350^{\circ}\text{C}$  and is condensed in the condenser at a pressure of  $75\text{KPa}$ . Determine the thermal efficiency of the cycle. (06 Marks)

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