Rajiv Gandhi University of Health Sciences, Karnataka Third Semester B. Pharm Degree Examination – JUNE-2019

Time: Three Hours Max. Marks: 75 Marks

Physical Pharmaceutics - I Q.P. CODE: 5010

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. State and explain Distribution law. Mention its limitations and applications.
- 2. Define Refractive Index. Discuss in detail working of Abbe's refractometer.
- 3. Define interfacial tension. Explain in detail any one method for the determination of the interfacial tension.

SHORT ESSAYS (Answer any Seven)

 $7 \times 5 = 35 Marks$

- 4. Explain ideal solution with examples.
- 5. Define dissociation constant. Write a note on its application in pharmacy.
- 6. Describe various methods for the determination of HLB value.
- 7. Write a note on surface free energy.
- 8. Classify organic molecular complexes with examples.
- 9. Describe any one method for analysis of complexes.
- 10. Describe in detail electrometric determination of pH.
- 11. Write a note on Henderson Hasselbalch equation.
- 12. Write a note on solubility of gases in liquids.

SHORT ANSWERS (Answer All)

 $10 \times 2 = 20 \text{ Marks}$

- 13. State Raoult's law.
- 14. Mention the applications of dipole moment in Pharmacy.
- 15. Write any two limitations of Langmuir adsorption isotherm.
- 16. Mention the applications of complexation in pharmacy.
- 17. What are chelates?
- 18. What are sandwich complexes?
- 19. What is pH Sorensen's scale?
- 20. What is buffer capacity?
- 21. Mention the applications of buffers in pharmacy.
- 22. What is polymorphism? Give any two examples.
