Rajiv Gandhi University of Health Sciences, Karnataka

First Semester B.Pharm Degree Examination - DEC-2017

Time: Three Hours Max. Marks: 75 Marks

Pharmaceutical Analysis - I Q.P. CODE: 5002

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. What are errors and classify them. Define accuracy and precision. Describe the steps to minimize errors.
- 2. What are Neutralization curves? Explain the selection of indicators in the titration between weak acid with strong base using neutralization curves.
- 3. Define oxidation and reduction. Explain the principle involved in titration with potassium dichromate. Give its applications with suitable examples.

SHORT ESSAYS (Answer any Seven)

 $7 \times 5 = 35 \text{ Marks}$

- 4. What is pharmaceutical analysis? Explain different types of analysis. What is its scope in pharmacy?
- 5. What are indicators? Explain the theory of indicators used in acid base titrations?
- 6. Explain the reason behind why water is not used in non-aqueous titration.
- 7. Give the mechanism of action of adsorption indicators with suitable examples.
- 8. Classify complexometric titrations. Explain each type with suitable examples.
- 9. What is gravimetry? Explain the following terms a) Digestion b) ignition c) Ash treatment d) Inceneration.
- 10. Write the principle, instrumentation and applications of conductometry.
- 11. Enumerate the various types of electrodes in potentiometry. Give the working of Calomel electrode.
- 12. What is polarography? Explain the terms a) limiting current b) polarographic maxima c) diffusion current d) supporting electrolytes.

SHORT ANSWERS 10 x 2 = 20 Marks

- 13. What are systematic errors and random errors? Give examples.
- 14. What is color change interval of an acid base indicator? Give its importance.
- 15. How do you prepare acetous Perchloric Acid?
- 16. "Complete and balance the equation: $KMnO_4+H2SO_4+H2C2O_4 \rightarrow "$
- 17. What is masking agent? Give example for masking by precipitation.
- 18. Write the conditions for process of digestion in Gravimetry.
- 19. What is Specific conductance and Molar conductance?
- 20. What is standard hydrogen electrode?
- 21. Define Qualitative Analysis and Quantitative Analysis.
- 22. What is Cerimetry? Give its applications.
