## Rajiv Gandhi University of Health Sciences, Karnataka

First Semester B.Pharm Degree Examination - Dec-2017

Time: Three Hours Max. Marks: 75 Marks

## Pharmaceutical Inorganic Chemistry Q.P. CODE: 5004

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

## LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$ 

- 1. Describe the various sources of impurities present in pharmaceutical substances.
- 2. Explain major buffers used in pharmaceutical preparations. What are the advantages and disadvantages of each?
- 3. Enlist in detail any one drug with their molecular formula, synonym (if any) method of preparation and use belongs to magnesium, aluminium and sodium containing antacid.

## **SHORT ESSAYS (Answer any Seven)**

 $7 \times 5 = 35 \text{ Marks}$ 

- 4. How do you carry out the limit test for chlorides in the given sample of potassium permanganate?
- 5. Define acids and basis according to various concepts.
- 6. Write the preparation and medicinal uses of calcium carbonate and sodium fluoride.
- 7. Give the method of preparation and uses of Aluminium hydroxide gel.
- 8. Describe the precautions for storage and handling of radioisotopes.
- 9. Write a note on physiological acid base balance.
- 10. Explain the method of preparation and assay of Ammonium chloride.
- 11. What are Haematinics? Write the preparation and assay of ferrous sulphate.
- 12. What are antimicrobials? Write a note on various iodine preparations.

SHORT ANSWERS  $10 \times 2 = 20 \text{ Marks}$ 

- 13. What is the role of thioglycollic acid in iron limit test?
- 14. State the meaning of the term opalescence and turbidity.
- 15. Ethanolic sulphate standard solution is used in limit test for sulphates. Give reason.
- 16. Define saline Cathartic. Give examples.
- 17. "Write the chemical reaction involved in the assay of Chlorinated lime?"
- 18. Write the molecular formula and medicinal uses of sodium thiosulphate.
- 19. Write a note on potash alum.
- 20. What is desensitizing agents? Give examples.
- 21. Define half life.
- 22. Define following terms i) Osmotic pressure ii) Isotonic solution.

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