Rajiv Gandhi University of Health Sciences, Karnataka

III Year Pharm-D Degree Examination – NOVEMBER 2015

Time: Three Hours Max. Marks: 70 Marks

PHARMACEUTICAL ANALYSIS

Q.P. CODE: 2862

Your answers should be specific to the questions asked Draw neat labeled diagrams wherever necessary

LONG ESSAYS (Answer any Two)

 $2 \times 10 = 20 \text{ Marks}$

- 1. Explain in detail the sampling techniques in IR spectroscopy.
- 2. Explain the principle, instrumentation and factors affecting fluorescence intensity.
- Write the principle and different types of conductometric titration curves with example.

SHORT ESSAYS (Answer any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. Explain with Graphical methods of end point determination in Potentiometry.
- 5. Define Electrophoresis and write a note on type of Electrophoresis.
- 6. State and explain the mathematical expression for Beer's and Lambert's Law?
- 7. Define Column Chromatography. Add a note on elution.
- 8. Explain the terms (a) HETP (b) Theoretical plate (c) Retention time (d) Retention volume.
- 9. Write a note on Flame Ionization Detector and Thermal conductivity detector.
- 10. Explain the concept of pre-derivatization & post-derivatization techniques in G.C. with relevant examples?
- 11. Explain the theory and applications of ion exchange chromatography.

SHORT ANSWERS $10 \times 2 = 20 \text{ Marks}$

- 12. Forbidden transitions.
- 13. Electromagnetic spectrum.
- 14. Nernst equation
- 15. Significance of guard columns in HPLC
- 16. Pharmaceutical applications of HPTLC
- 17. What is the difference between silica gel, H, G and GF?
- 18. Give a specific spray reagent to detect the following compounds by TLC a) Sulphanilamide b) Amino acids c) Alkaloid d) Phenols
- 19. R_f and R_x values and their significance
- 20. Finger print range and its significance.
- 21. Applications of atomic absorption spectroscopy.
