

# GBGS SCHEME

15BT51

(08 Marks)

## Fifth Semester B.E. Degree Examination, Jan./Feb.2021 **Bio Kinetics & Bio Reaction Engineering**

Time: 3 hrs. Max. Marks: 80 2. Any revealing of identification, appeal to evaluator and /or equations written eg, 42+8 = 50, will be treated as malpractice. Note: Answer any FIVE full questions, choosing ONE full question from each module. Module-1 a. Define the following with example: (i) Rate (ii) Order (iii) Mechanism Important Note: 1. On completing your answers, compulsorily draw diagonal cross lines on the remaining blank pages (iv) Elementary and Non Elementary reactions. (08 Marks) Explain the temperature dependency of rate constant given by collision theory. (08 Marks) Explain Integral and differential method of analyzing kinetic data. (08 Marks) Derive equation for II order biomolecular reaction for  $M \neq 1$ . (08 Marks) Module-2 Derive a performance equation for batch reactor. 3 (08 Marks) In isothermal batch reactor, the conversion of liquid reactant A achieved in 13 min is 70%. Find the space time and space velocity to effect this conversion in plug flow reactor and in mixed flow reactor. Consider first order kinetics. (08 Marks) OR Explain the qualitative discussion of product distribution in parallel reaction. (08 Marks) Explain the size comparison of CSTR and PFR. (08 Marks) Module-3 What are the main cause for non ideality in a vessel? Explain. (04 Marks) Explain RTD curve. (06 Marks) Derive the relationship between C, E and F curve. (06 Marks) OR Derive the RTD equation for PFR. (08 Marks)

Explain in detail the conversion for I<sup>st</sup> order reactions.

7 Derive M-M equation starting from all assumptions. (10 Marks) Explain compitative inhibition. b. (06 Marks) 1 of 2

#### OR

8 a. The following data is obtained for two different initial enzyme concentration for an enzyme catalysed reaction.

$V([E_0] = 0.015 \mathrm{g/l})$	[S]	$V([E_0] = 0.000875 g/l)$
g / <i>l</i> – min	g/l	g / <i>l</i> – min
1.14	20.0	0.67
0.87	10.0	0.51
0.70	6.7	0.41
0.59	5.0	0.34
0.50	4.0	0.29
0.44	3.3	A. J.
0.39	2.9	
0.35	2.5	¥

- (i) Find Km.
- (ii) Find Vm for  $[E_0] = 0.015 \text{ g} / l$
- (iii) Find Vm for  $[E_0] = 0.00875 \text{ g} / l$
- (iv) Find K<sub>2</sub>

(10 Marks)

b. Explain allosteric and feedback regulation.

(06 Marks)

### Module-5

9 a. Derive Monad growth kinetics using ideal reactor.

(06 Marks)

b. Explain in detail primary and secondary product formation kinetics.

(10 Marks)

#### OR

- 10 a. Define the following:
  - (i) Degree of reduction.
  - (ii) Overall growth yield.
  - (iii) Respiratory quotient.

(06 Marks)

- b. With neat sketch, explain the following:
  - (i) Sterilization of liquid media.
  - (ii) Filter sterilization.

(10 Marks)