CBCS SCHEME

USN 21BT51

Fifth Semester B.E. Degree Examination, Dec.2023/Jan.2024 Biokinetics and Bioreaction Engineering

Time: 3 hrs. Max. Marks: 100

Note: Answer any FIVE full questions, choosing ONE full question from each module.

Module-1

- 1 a. Explain briefly about Arrhenius and Collision theory. (10 Marks)
 - b. The rate of reaction is doubled for each 10°C rise. Derive the correlation between activation energy and T. Also find the value if E at 27°C and 327°C. (10 Marks)

OR

- 2 a. Briefly explain Differential method of analysis and Integral method of analysis. (10 Marks)
 - b. Derive an integrated rate equation for a unimolecular First order reaction both in terms of concentration and conversion. (10 Marks)

Module-2

- 3 a. Derive the performance equation for both reactor for constant volume system interms of concentration and conversion. (10 Marks)
 - b. Enzyme E catalyses the fermentation of substrate A to produce R. Find the size of mixed flow reactor needed for 95% conversion of reactant in a feed stream of 25ℓpm of reactant (2 moles/lit) and enzyme. The kinetics of fermentation at this enzyme concentration is given as A → R.

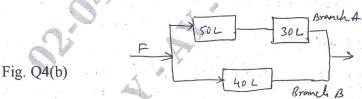
$$-r_A = \frac{0.1C_A}{1 + 0.5C_A}$$
 moles/liter.min

(10 Marks)

OR

- 4 a. Discuss the various reasons for non ideality in reactor operations. Explain the experimental method to determine the residence time distribution and list out the properties of tracer.

 (10 Marks)
 - b. The reactor set up shown in Fig. Q4(b) consists of 3 plug glow reactors in 2 parallel batches. Brach A has a reactor volume of 50 liters, followed by a reactor volume of 30 liters. Branch B has a reactor of volume 40 litres. What fraction of the feed should go to the branch A?



Module-3

- 5 a. Derive Michaelis Menten equation, stating with all assumptions. (10 Marks
 - Explain the Wineweer Burke plot, Eadie Hofstee and Hanes wolf plot to estimate the M – M parameters.
 (10 Marks)

6 a. Beef Wiver catalyst has been used to accelerate the decomposition of H₂O₂ to yield H₂O and O₂. The concentration of H₂O₂ is given as a function of time for a reaction mixture with a P^H of 6.7 and maintained at 30°C.

-	t, min	0	10	20	50	100
	Concentration of H ₂ O ₂ , mol/lit	0.02	0.0175	0.0158	0.0106	0.005

Determine M -M parameters, Umax and Km.

(10 Marks)

- b. Explain the following terms:
 - i) Competitive inhibition
- ii) Non competitive inhibition
- iii) Allosteric enzyme iv) Free and Bound enzyme.

(10 Marks)

Module-4

7 a. With a neat graphical representation, explain the Monod model.

(10 Marks)

b. Explain Batch growth kinetics.

(10 Marks)

OR

- 8 a. Explain Elemental balance for a biological reactions in which no extra cellular products produced. (10 Marks)
 - b. Aerobic degradation of benzoic acid by a mixed culture of microbes can be represented by the following reaction: C₆H₅COOH + aO₂ + bNH₃ → cC₅H₇NO₂ + dH₂O + eCO₂.
 Determine i) a, b, c, d and e if respiratory quotient (RQ) is 0.9.
 - ii) Yield coefficient $Y_{x/}$, $Y_{x/}$

(10 Marks)

Module-5

9 a. Discuss briefly media required for Fermentation process.

(10 Marks)

b. Explain Thermal death kinetics of microorganism.

(10 Marks)

OR

10 a. Differentiate between Batch and Continuous Sterilization of Liquid media.

(10 Marks)

b. With a neat sketch, explain Filter Sterilization.

(10 Marks)