Rajiv Gandhi University of Health Sciences, Karnataka

II Year B.Sc. (MLT) Degree Examination - 21-Nov-2024

Time: Three Hours Max. Marks: 80 Marks

BIOCHEMISTRY - PAPER II (RS3) Q.P. CODE: 3156

Your answers should be specific to the questions asked. Draw neat, labeled diagrams wherever necessary.

LONG ESSAY (Any Two)

 $2 \times 10 = 20 \text{ Marks}$

- Define enzymes. Explain different types of enzyme inhibition with example.
- 2. Define Beer-lamberts law. Draw a neat diagram and explain the parts and applications of colorimeter.
- 3. Explain the regulation of blood glucose level in body. Add a note on glycated HbA1c.

SHORT ESSAY (Any Six)

 $6 \times 5 = 30 \text{ Marks}$

- 4. What is the difference between normal and molar solution? What is the normality of 1000 ml of 2M Sulfuric acid?
- 5. Creatinine clearance test.
- 6. Transamination reaction and its importance.
- 7. Explain how starch is digested.
- 8. Functions and deficiency manifestations of Niacin.
- 9. Principle and applications of chromatography.
- 10. Secondary structure of proteins.
- 11. Collection of venous sample for fasting plasma glucose estimation.

SHORT ANWERS (Any Ten)

 $10 \times 3 = 30 \text{ Marks}$

- 12. Pleural fluid.
- 13. Write normal levels of Serum Urea, Serum Creatinine and Fasting plasma glucose.
- 14. What is gout? Enumerate two causes of gout.
- 15. Name any three Urine preservatives.
- 16. Name any three normal constituents of urine. Name one test each used to detect them.
- 17. What are essential amino acids? Give two examples.
- 18. Preparation of calibration curve.
- 19. Name the defective enzyme in a) Maple syrup urine disease b) Alkaptonuria c) Phenyl Ketonuria.
- 20. Functions of Cobalamin.
- 21. Invert sugar.
- 22. Difference between DNA and RNA.
- 23. Principle of Freezing Point Depression of Osmometry.
